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#### Key indicators

Single-crystal X-ray study T = 292 KMean  $\sigma(\text{C}-\text{C}) = 0.003 \text{ Å}$  R factor = 0.058 wR factor = 0.127 Data-to-parameter ratio = 17.0

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# 2,3-Bis(hydroxymethyl)-6-isopropylperhydro-2,3,4a,6,7a-pentaazacyclopenta[cd]indene-1,4-dione

The molecule of the title compound,  $C_{11}H_{19}N_5O_4$ , contains three fused rings, namely two nearly planar imidazole rings and one non-planar triazine ring. The latter ring displays a chair conformation. Two hydroxylmethylene groups are linked to two N atoms from separate rings of the glycoluril system.

### Comment

Glycoluril derivatives have applications in many fields such as explosives, slow-release fertilizers, cross-linkers, iodogen stabilizers of organic compounds against photodegradation and reagents in combinatorial chemistry (Wu *et al.*, 2002). As a part of our ongoing investigation of glycoluril derivatives (Li *et al.*, 2005), we present here the structure of the title compound, (I).



The molecular structure of (I) is shown in Fig. 1. Within the nearly planar five-membered ring, the N-C(carbonyl) bond



#### Figure 1

The molecular structure of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are represented by circles of arbitrary radius.

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distances are much shorter than the other N-C distances (Table 1), indicating electron delocalization. The sixmembered ring displays a chair conformation.

The crystal structure is stabilized by inter- and intramolecular  $O-H\cdots O$  and  $C-H\cdots O$  hydrogen-bonding interactions (Table 2).

## Experimental

Isopropylamine (0.58 g, 10 mmol) and formaldehyde (2.4 g, 40 mmol) were added to a stirred solution of tetrahydroimidazo[4,5-d]imidazole-2,5-dione (0.71 g, 5 mmol) in acetonitrile (50 ml) under a nitrogen atmosphere. The mixture was stirred overnight at room temperature. The solvent was evaporated to dryness and the compound was purified by column chromatography to obtain the title compound (yield 0.28 g, 20%) as a colorless soid. Crystals of (I) suitable for data collection were obtained by slow evaporation of a methanol solution at 283 K.

#### Crystal data

 $\begin{array}{l} C_{11}H_{19}N_5O_4\\ M_r = 285.31\\ \text{Orthorhombic, }Pbca\\ a = 12.778 \ (3) \ \text{\AA}\\ b = 12.655 \ (3) \ \text{\AA}\\ c = 15.926 \ (4) \ \text{\AA}\\ V = 2575.2 \ (11) \ \text{\AA}^3 \end{array}$ 

#### Data collection

Bruker SMART 4K CCD areadetector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: none 28542 measured reflections

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.058$   $wR(F^2) = 0.127$  S = 0.833154 reflections 185 parameters Z = 8  $D_x = 1.472 \text{ Mg m}^{-3}$ Mo K\alpha radiation  $\mu = 0.11 \text{ mm}^{-1}$  T = 292 (2) KPlate, colorless  $0.30 \times 0.20 \times 0.04 \text{ mm}$ 

3154 independent reflections 1428 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.143$  $\theta_{\text{max}} = 28.3^{\circ}$ 

H-atom parameters constrained  $w = 1/[\sigma^2(F_o^2) + (0.0506P)^2]$ where  $P = (F_o^2 + 2F_c^2)/3$   $(\Delta/\sigma)_{max} < 0.001$   $\Delta\rho_{max} = 0.18 \text{ e} \text{ Å}^{-3}$  $\Delta\rho_{min} = -0.23 \text{ e} \text{ Å}^{-3}$ 

#### Table 1

Selected geometric parameters (Å, °).

C3-N1	1.486 (3)	C7-N5	1.367 (3)
C4-N1	1.454 (3)	C7-N3	1.370 (3)
C4-N2	1.461 (3)	C8-N2	1.451 (3)
C6-N2	1.365 (3)	C8-N3	1.452 (3)
C6-N4	1.373 (3)		
N2-C6-N4	107.7 (2)	N5-C7-N3	109.3 (2)

able	2	

Т

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D{\cdots}A$	$D - \mathbf{H} \cdots A$
$C5-H5B\cdots O2$	0.97	2.59	2.925 (3)	101
O4−H4···O3	0.82	1.92	2.728 (3)	168
$C10-H10B\cdots O4^{i}$	0.97	2.58	3.494 (3)	157
$O3-H3A\cdots O2^{i}$	0.82	1.90	2.713 (2)	173

Symmetry code: (i)  $x - \frac{1}{2}, -y + \frac{1}{2}, -z$ .

All H atoms were positioned geometrically and treated as riding, with C-H = 0.96 (CH<sub>3</sub>), 0.97 (CH<sub>2</sub>) or 0.98 Å (CH) and O-H = 0.82 Å, and with  $U_{iso}$ (H) =  $1.2U_{eq}$ (C) or  $1.5U_{eq}$ (methyl C,O).

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

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